Benzene Substituents: Reactions, Concepts, and Synthesis Practice

Hey, gang! Welcome to the worksheet for the benzene substituent chemistry unit. Although there are a bunch of videos in this section, this one worksheet contains all the practice. It might be a lot, but I know you all are up to the task.

We'll cover everything from acid-base properties of phenols to cope rearrangements and back. If anything is confusing, check out the solutions and/or solutions walkthrough video ②!

- **1.)** To get this party started, let's hit some acid-base & concept questions:
- a.) Given the following three molecules, rank them 1-3 with 1 being the **weakest** acid and 3 being the **strongest** acid.

b.) Given the following three molecules, rank them 1-3 with 1 being the weakest base and 3 being the strongest base.

c.) Given the following three molecules, rank them 1 – 3 with 1 being the **least** susceptible and 3 being the **most susceptible** to nucleophilic aromatic substitution.

$$O_2N$$
 O_2N
 O_2N

d.) Okay, let's shift away from ranking.

In this next problem, circle the reaction A, B, or C that will yield the product molecule (shown below) in the highest yield:

A
$$H_3CO$$

ONA + CI

NO2

NO2

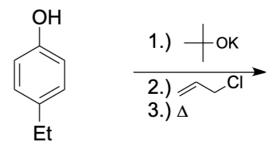
NO2

NO2

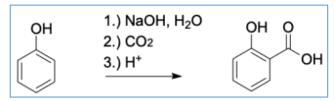
NO2

HNO3, H_2SO_4

2.) Okay—enough of looking at and ranking reactions, let's complete some of our own! For the reactions below, you know the drill: either complete the reaction by predicting the major organic product, providing the reactant(s), or providing the reagent(s).



- **3.)** Now that we've completed *plenty* of reactions, let's transition to mechanism mode.
- a.) Draw the full arrow pushing mechanism for the reaction displayed below



b.) Justify the observed reaction below with a mechanism

Okay, gang, we're close to the end. After this synthesis question below there's one more question, which is more of a bonus.

4.) Using benzene and ethanol as your only carbon source, synthesize the target molecule shown below.

EtOH

5.) BONUS QUESTION: Draw a mechanism to illustrate the reaction displayed below.

$$\begin{array}{c|c} & \Delta & & \\ \hline \\ O & \\ \hline \end{array}$$